amphoteric

A chemical species that behaves both as an acid and as a base is called amphoteric. This property depends upon the medium in which the species is investigated: $\text{H}_2\text{SO}_4$ is an acid when studied in water, but becomes amphoteric in superacids.

Source:
PAC, 1994, 66, 1077 (Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)) on page 1083