**exergonic (exoergic) reaction**

This expression is often applied to a reaction for which the overall standard Gibbs energy change $\Delta G^\circ$ is negative. Some workers use this term with reference to a negative value of $\Delta H^\circ$ at the absolute zero of temperature.

**Source:**
PAC, 1996, 68, 149 (A glossary of terms used in chemical kinetics, including reaction dynamics (IUPAC Recommendations 1996)) on page 165