

fulminates

1. Compounds having the structure $\text{RON}=\text{C}:$. So called because fulminic acid (actually $\text{HC}\equiv\text{N}^+-\text{O}^-$ formonitrile oxide) was previously considered to be $\text{HON}=\text{C}:$.
2. Salts of fulminic acid, e.g. $\text{Na}^+[-\text{C}\equiv\text{N}^+\text{O}^-]$.

Source:

PAC, 1995, 67, 1307 (*Glossary of class names of organic compounds and reactivity intermediates based on structure (IUPAC Recommendations 1995)*) on page 1336